SYNTHESES OF NEW TXA₂/PGH₂-RECEPTOR ANTAGONISTS AND THEIR BIOLOGICAL PROPERTIES

P. Deicke and U. Klar*
Research Laboratories of Schering AG,
Müllerstr. 170-178, W-1000 Berlin 65, FRG

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SUMMARY: The syntheses of new types of TXA₂/PGH₂-receptor antagonists and some structure activity relationships are discussed. It is shown that a suitable substituent attached directly to the carbon bridge of the 2-oxabicyclo[2.2.1]heptane system of the standard TXA₂/PGH₂-receptor agonist U 46619 changes the biological profile from an agonist to an antagonist and results in biologically very active compounds.

INTRODUCTION AND CONCEPT: Thromboxane A₂ (TXA₂) and Prostaglandin H₂ (PGH₂) are produced in vivo from arachidonic acid via the 11-cyclooxygenase pathway¹. Both compounds are potent stimulators of platelet aggregation, induce contraction of vascular and airways smooth muscle, and are considered to interact with the same cell-surface receptor or receptor-type². Because of the association of TXA₂/PGH₂ with a number of pathological states³, e.g., myocardial ischemia, coronary vasospasm, asthma, and hepatorenal diseases, the intensive search for specific TXA₂/PGH₂-receptor antagonists led to the synthesis of a variety of prostanoid (e.g. SQ 29548, BMS-180291, S-145, Ono-8809, ICI-192605, GR-32191, EP 092) and non-prostanoid structures (e.g. BM 13177, 13505, Bay-u-3405, L670596)⁴.

In one of our own approaches, we tried to change the agonist profile of the stable TXA₂/PGH₂-mimic U 46619⁵ (I, Z=H, Scheme 1) adding a residue Z to the carbon bridge of the bicyclic ring system to obtain compounds with antagonist properties. Our synthetic strategy is outlined in the retrosynthetic analysis (Scheme 1).

Scheme 1

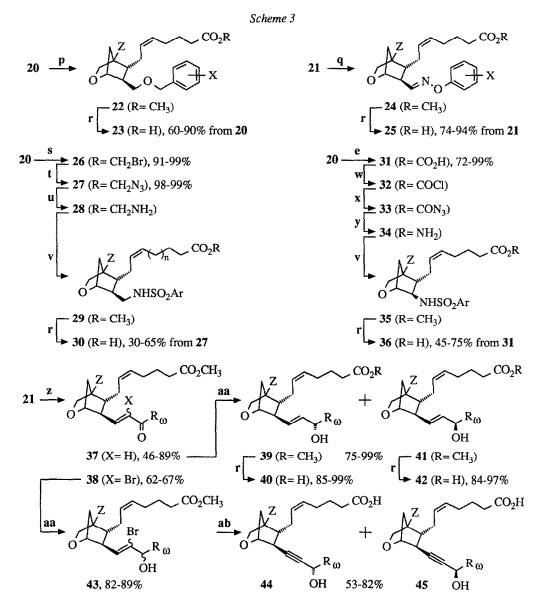
$$CO_2H$$
 CO_2H
 OH
 OH

Removal of the ω -chain in I and opening of the cyclic ether moiety leads to the prostanoid precursor III. Cleavage of the double bond results in the Baeyer-Villiger product derived from ketone V, in which the residue Z is introduced by 1,4-addition to the key intermediate VI. VI finally can be prepared regions electively from Corey-lactone VII, which is available in both enantiomeric forms.

SYNTHESIS: The synthesis started from (+)- as well as (-)- Corey-lactone 1, although it is shown here for (-)-1 with the correct absolute configuration of the natural prostanoids (Scheme 2). The hydroxy groups were protected as tert.-butyldiphenylsilyl (TBDPS) and tetrahydropyranoyl (THP) ether, respectively, by standard methods. The formal exchange of the lactone oxygen by a carbon atom was achieved in a three step sequence according to P.A. Aristoff et al.⁶ by addition of deprotonated dimethyl methylphosphonate, Jones-oxydation to 1,4-diketone 6 and cyclisation to the key intermediate 7. The residue Z was introduced either by Cu(I) catalyzed 1,4-addition of Grignard-reagents when Z represents an alkyl or aryl group, or by Ni(I) catalyzed 1,4-addition of diethylaluminum alkynes in the cases were Z is a terminal alkyne⁷. Baeyer-Villiger oxidation gave a nearly 1:1 mixture of regioisomers 9 and 11, which could be separated by chromatography on silica gel after reduction to the lactols 13 and 14. Reoxidation to the corresponding lactones 9 or 11 and removal of the THP-ether gave compounds 10 and 12 in which the regioisomers could be readily assigned based on the coupling pattern (AB versus ABX) of the methylene protons adjacent to the lactone oxygen in the ¹H-NMR spectrum. The introduction of the α-chain into 13 and 14 by Wittig-reaction, formation of the methyl ester and selective cleavage of the THP-ether led to the 1,4-diols 16 and 18. Although the separation of the regioisomers could be performed much more readily with diols 16 and 18 independent of the nature of Z, Mitsunobu cyclisation⁸ of this mixture yielded only the desired ether 19, while 16 gave mainly other, not identified products. Alcohol 20, obtained after removal of the silvl group, is the key intermediate for the introduction of various ω -chains. As indicated in Scheme 3, the alcohol was directly transformed into benzylic ethers (23), sulfonamides (30, 36), or -after oxidation to the aldehyde 21into oxime ethers (25) or prostanoid chains via Wittig-Horner reaction. After in situ bromination of deprotonated phosphonate with NBS, the resulting vinyl bromide 38 was reduced to give then on HBr elimination the 13,14-triple bond. The epimeric alcohols were readily separated by chromatography on silica gel. Their configuration at C-15 was assigned on the assumption that the more active epimers possess the same 15α-configuration of the hydroxyl group as the natural prostaglandins.

RESULTS: The selected compounds listed in *Tables 1* to 4 allow some general statements for structure-activity relationships in this structural class. As depicted in *Table 1*, the TXA₂-agonist properties for the parent compound U 46619 (entry 1b) can be converted into TXA₂-antagonist properties if Z is larger than methyl. Furthermore, on inverting the absolute configuration as shown in *Table 2* for different types of ω -chains and some selected groups Z, the affinity to the TXA₂/PGH₂-receptor is improved significantly. The effect of the substitution pattern for Z=phenyl on the receptor binding for different types of ω -chains is given in *Table 3*. Compounds with Z=biphenyl and sulfonamides as ω -chains show the strongest binding. The introduction of an additional methylene group between the bicyclic ring system and the sulfonamide nitrogen has no significant influence on the receptor binding (entries 8, *Table 3* and 9, *Table 4*), but the ability to inhibit platelet aggregation, however, is enhanced 10-fold. *Table 4* shows the influence of the length of the α -chain as well as the double bond on receptor binding for different substituted aryl sulfonamides. It is obvious that a double bond is not necessary and that the shortening of the α -chain by one methylene group is well tolerated and in some cases even improves receptor affinity. All derivatives listed in *Table 4* are pure TXA₂/PGH₂-antagonists. For example, compound of entry 3 is 100-fold more active in inhibiting the aggregation of washed human platelets than the standard TXA₂-antagonist SQ 29548.

Scheme 2

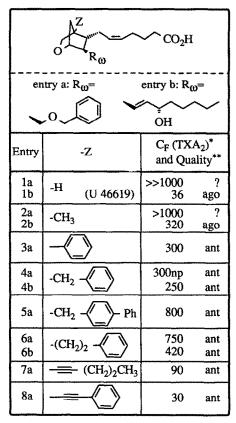


CONDITIONS: a: TBDPS-Cl, imidazole, DMF, 23°C, 2h, argon; b: K₂CO₃, MeOH, 23°C, 16h, argon, SiO₂; c: dihydropyrane, cat. p-TsOH, CH₂Cl₂, 23°C, 1h, argon, SiO₂; d: dimethyl methylphosphonate, n-BuLi, THF, -78°C, 1.5h, argon, SiO₂; e: Jones-oxidation f: K₂CO₃, 18-crown-6, toluene, 23°C, 16h, argon, SiO₂; g: Z-Mg-X, cat. CuOAc, ether, -15°C, 0.5-3h or Z-AlEt₂, cat. Ni(I), ether, -10°C, argon, SiO₂; h: m-CPBA, NaHCO₃, CH₂Cl₂, 23°C, 10-18h, argon, SiO₂; i: cat. PPTs, EtOH, 55°C, 2-7h, argon, SiO₂; k: DIBAL, toluene, -70°C, 0.5-1h, argon; l: n-BuLi, HMDS, carboxybutyltriphenylphosphonium bromide (n=1) or carboxypropyltriphenylphosphonium bromide (n=0), THF, 30-60°C, 2-6h; CH₂N₂, ether, CH₂Cl₂, 5°C, 0.5-2h, argon, SiO₂; m: PPh₃, DEAD, toluene, 50-60°C, 3-20h, argon, SiO₂; n: 1M TBAF, THF, 23°C,

1-4h, argon, SiO₂; o: Swern-oxidation; p: Hal-CH₂-C₆H₄-X. 50% KOH, cat. Bu₄NHSO₄, 23°C, 16h, argon, SiO₂; q: H₂N-O-C₆H₄-X, EtOH, cat. pyridine, 50°C, 2-3h, argon, SiO₂; r: 5% LiOH, MeOH, 23°C, 5-16h, argon, SiO₂; s: CBr₄, Ph₃P, collidine, acetonitrile, 23°C, 16h, argon, SiO₂; t: NaN₃, DMF, 60°C, 2-4h, SiO₂; u: Ph₃P, THF, 23°C, 16h, H₂O, rf, 1h; v: CISO₂Ar, CH₂Cl₂ 23°C, 0.5h, argon, SiO₂; w: SOCl₂, CH₂Cl₂, 5-23°C, 6h, argon; x: NaN₃, H₂O, CH₂Cl₂, cat. Bu₄NHSO₄, 5°C, 1h, argon; y: F₃CCO₂H, H₂O, CH₂Cl₂, 50°C, 4h, argon, SiO₂; z: NaH, phosphonate, (NBS), DME, or LiCl, phosphonate, DBU, acetonitrile, 23°C, 3-23h, argon, SiO₂; aa: NaBH₄, MeOH, -40°C, 1h, argon, SiO₂; ab: KO-t-Bu, DMSO, THF, 23°C, 2-4h, argon, SiO₂.

 $\begin{tabular}{l} Table 1 \\ Influence of group Z on receptor affinity \\ and TXA_2-agonist/antagonist properties \\ \end{tabular}$

Table 2
Influence of absolute configuration on receptor affinity and TXA₂-agonist/antagonist properties



entry nat: CO_2H entry ent: CO_2H							
Entry	-Z	-R _∞	ar	XA ₂) nd ality			
1-nat 1-ent	-Ph		300 60	ant ant			
2-nat 2-ent	-≅-Ph	$\sqrt{\circ}$	30 12	ant ant			
3-nat 3-ent	-{\$}F		125 60	ant ant			
4-nat 4-ent		√0 CN	900 6	? pa			
5-nat 5-ent		✓✓✓✓ OH	650 25	? ant			
6-nat 6-ent		✓ OH OF F	375 2	ant pa			

The TXA_2/PGH_2 -receptor affinities are given as competition factor $C_F = (ID_{50}\text{-test})/(ID_{50}\text{-standard})$; SQ 29548 is used as standard TXA_2/PGH_2 -receptor antagonist; np means not parallel.

The quality of compounds binding to the TXA₂/PGH₂-receptor is defined as follows: antagonists are able to inhibit the aggregation of (human) platelets as well as the contraction of rabbit aorta induced by the standard TXA₂/PGH₂-agonist U 46619 (1·10⁻⁶M); partial agonists either amplify ADP induced platelet aggregation and/or contract rabbit aorta but still inhibit platelet aggregation induced by U 46619; agonists aggregate platelets and/or contract the rabbit aorta by their own.

Table 3:
Influence of phenyl substitution on receptor affinity and TXA₂-agonist/antagonist properties

CO₂H R_{ω} $C_F(TXA_2)$ Entry X $-R_{\omega}$ and Quality 3-F 250 ant 1 4-F 2 120 ant 3 4-OH 30 ant 4-Ph 4 Ĥ 2np ant 5 3-F 120 ant 170np 6 4-F ant 7 4-OH 520 ant 8 4-Ph 2np Ĥ ant 9 3-F 10 ant 10 4-F 16 ant 11 4-Ph 16np ant 12 4-F 26 ant 20np pa 13 4-Ph OH 14 4-F 4 ago 4-Ph 15 12np pa OH

Table 4: Influence of double bond and α-chain length on receptor affinity of TXA₂-antagonists

$O \xrightarrow{\longrightarrow}_{n} CO_{2}H$ $O \xrightarrow{\longrightarrow}_{NHSO_{2}Ar}$						
Entry	n	Δ	-Ar	C _F (TXA ₂)		
1 2 3 4	1 1 0 0	Z sat Z sat	—⟨¯> cı	0.4 np 1.4 0.1 np 0.3 np		
5 6 7 8	1 1 0 0	Z sat Z sat	CI CI	4.2 np 12.0 6.0 np 5.0		
9 10 11	1 0 0	Z Z sat	———— CH₃	4.0 np 0.05 0.1		
12 13	0	Z sat	─ F	0.05 0.1		

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